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## Structure Reports

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## Methyl 2-[2-(6-chloropyrimidin-4-yl-oxy)phenyl]-3,3-dimethoxypropanoate

Chao Sheng, Qing-Bing Xu, Yuan-Yuan Liu and Hong-Jun Zhu\*

Department of Applied Chemistry, College of Science, Nanjing University of Technology, Nanjing 210009, People's Republic of China  
Correspondence e-mail: zhu hj@njut.edu.cn

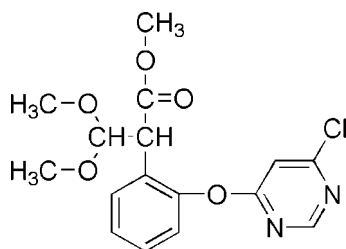
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Key indicators: single-crystal X-ray study;  $T = 294$  K; mean  $\sigma(\text{C}-\text{C}) = 0.008$  Å;  $R$  factor = 0.078;  $wR$  factor = 0.173; data-to-parameter ratio = 14.9.

In the title compound,  $\text{C}_{16}\text{H}_{17}\text{ClN}_2\text{O}_5$ , the dihedral angle between the aromatic rings is  $77.36$  (4)°. An intramolecular  $\text{C}-\text{H}\cdots\text{O}$  interaction results in the formation of a planar [r.m.s. deviation =  $0.103$  (2) Å] five-membered ring, which is oriented at a dihedral angle of  $4.84$  (4)° with respect to the adjacent benzene ring. In the crystal structure, weak intermolecular  $\text{C}-\text{H}\cdots\pi$  interactions are found.

## Related literature

For a related structure, see: Bowden & Brown (1996). For bond-length data, see: Allen *et al.* (1987).



## Experimental

## Crystal data

$\text{C}_{16}\text{H}_{17}\text{ClN}_2\text{O}_5$   
 $M_r = 352.77$   
Triclinic,  $P\bar{1}$   
 $a = 9.5030$  (19) Å

$b = 10.051$  (2) Å  
 $c = 11.162$  (2) Å  
 $\alpha = 101.24$  (3)°  
 $\beta = 108.47$  (3)°

$\gamma = 113.42$  (3)°  
 $V = 862.6$  (5) Å<sup>3</sup>  
 $Z = 2$   
Mo  $K\alpha$  radiation

$\mu = 0.25$  mm<sup>-1</sup>  
 $T = 294$  K  
 $0.20 \times 0.20 \times 0.05$  mm

## Data collection

Enraf–Nonius CAD-4 diffractometer  
Absorption correction:  $\psi$  scan (North *et al.*, 1968)  
 $T_{\min} = 0.952$ ,  $T_{\max} = 0.988$   
3346 measured reflections

3140 independent reflections  
1427 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.035$   
3 standard reflections  
frequency: 120 min  
intensity decay: 1%

## Refinement

$R[F^2 > 2\sigma(F^2)] = 0.078$   
 $wR(F^2) = 0.173$   
 $S = 1.07$   
3140 reflections

211 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\text{max}} = 0.43$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.45$  e Å<sup>-3</sup>

**Table 1**  
Hydrogen-bond geometry (Å, °).

| $D-\text{H}\cdots A$                       | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--|--------------|--------------------|-------------|----------------------|
| $\text{C6}-\text{H6A}\cdots\text{O5}$      | 0.98         | 2.25               | 2.777 (6)   | 113                  |
| $\text{C1}-\text{H1B}\cdots\text{Cg2}^i$   | 0.96         | 2.97               | 3.696 (5)   | 134                  |
| $\text{C16}-\text{H16A}\cdots\text{Cg1}^i$ | 0.93         | 2.85               | 3.661 (4)   | 146                  |

Symmetry code: (i)  $-x + 1, -y + 1, -z$ .  $\text{Cg1}$  and  $\text{Cg2}$  are centroids of the  $\text{C7}-\text{C12}$  and  $\text{N1}/\text{N2}/\text{C13}-\text{C16}$  rings, respectively.

Data collection: *CAD-4 Software* (Enraf–Nonius, 1985); cell refinement: *CAD-4 Software*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL* (Sheldrick, 2008); software used to prepare material for publication: *SHELXTL* and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: HK2749).

## References

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**supplementary materials**

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## Methyl 2-[2-(6-chloropyrimidin-4-yloxy)phenyl]-3,3-dimethoxypropanoate

C. Sheng, Q.-B. Xu, Y.-Y. Liu and H.-J. Zhu

### Comment

The title compound can be used as an intermediate in the preparation of azoxystrobin, which is an important fungicide (Bowden & Brown, 1996). We report herein the crystal structure of the title compound, which is of interest to us in the field.

In the molecule of the title compound (Fig. 1), the bond lengths (Allen *et al.*, 1987) and angles are within normal ranges. Rings A (C7-C12) and B (N1/N2/C13-C16) are, of course, planar and the dihedral angle between them is  $A/B = 77.36(4)^\circ$ . Intramolecular C-H $\cdots$ O interaction (Table 1) results in the formation of a planar five-membered ring C (O5/C6-C8/H6A), which is oriented with respect to the adjacent ring A at a dihedral angle of  $A/C = 4.84(4)^\circ$ .

In the crystal structure, weak C—H $\cdots$  $\pi$  interactions (Table 1) are found.

### Experimental

The title compound was prepared according to a literature method (Bowden & Brown, 1996). Crystals suitable for X-ray analysis were obtained by dissolving the title compound in methanol and evaporating the solvent slowly at room temperature for 8 d.

### Refinement

H atoms were positioned geometrically with C-H = 0.93, 0.98 and 0.96 Å for aromatic, methine and methyl H atoms, respectively, and constrained to ride on their parent atoms, with  $U_{\text{iso}}(\text{H}) = xU_{\text{eq}}(\text{C})$ , where  $x = 1.5$  for methyl H and  $x = 1.2$  for all other H atoms.

### Figures

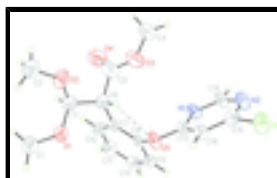


Fig. 1. The molecular structure of the title molecule with the atom-numbering scheme. Displacement ellipsoids are drawn at the 50% probability level. Hydrogen bond is shown as dashed line.

## Methyl 2-[2-(6-chloropyrimidin-4-yloxy)phenyl]-3,3-dimethoxypropanoate

### Crystal data

$\text{C}_{16}\text{H}_{17}\text{ClN}_2\text{O}_5$

$M_r = 352.77$

Triclinic,  $P\bar{1}$

$Z = 2$

$F_{000} = 368$

$D_x = 1.358 \text{ Mg m}^{-3}$

# supplementary materials

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Hall symbol: -P 1  
 $a = 9.5030$  (19) Å  
 $b = 10.051$  (2) Å  
 $c = 11.162$  (2) Å  
 $\alpha = 101.24$  (3)°  
 $\beta = 108.47$  (3)°  
 $\gamma = 113.42$  (3)°  
 $V = 862.6$  (5) Å<sup>3</sup>

Mo  $K\alpha$  radiation,  $\lambda = 0.71073$  Å  
Cell parameters from 25 reflections  
 $\theta = 9\text{--}12^\circ$   
 $\mu = 0.25$  mm<sup>-1</sup>  
 $T = 294$  K  
Needle, colorless  
 $0.20 \times 0.20 \times 0.05$  mm

## Data collection

Enraf–Nonius CAD-4  
diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 294$  K

$\omega/2\theta$  scans

Absorption correction:  $\psi$  scan  
(North *et al.*, 1968)

$T_{\min} = 0.952$ ,  $T_{\max} = 0.988$

3346 measured reflections

3140 independent reflections

1427 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.035$

$\theta_{\max} = 25.3^\circ$

$\theta_{\min} = 2.1^\circ$

$h = 0 \rightarrow 11$

$k = -12 \rightarrow 11$

$l = -13 \rightarrow 12$

3 standard reflections

every 120 min

intensity decay: 1%

## Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.078$

$wR(F^2) = 0.173$

$S = 1.07$

3140 reflections

211 parameters

Primary atom site location: structure-invariant direct  
methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring  
sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.060P)^2]$$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\max} < 0.001$

$\Delta\rho_{\max} = 0.43$  e Å<sup>-3</sup>

$\Delta\rho_{\min} = -0.45$  e Å<sup>-3</sup>

Extinction correction: none

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > \sigma(F^2)$  is used only for calculating  $R$ -

factors(gt) etc. and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ - factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

|      | $x$          | $y$          | $z$          | $U_{\text{iso}}^*/U_{\text{eq}}$ |
|------|--------------|--------------|--------------|----------------------------------|
| C1   | 0.16863 (16) | 0.54848 (17) | 0.23573 (15) | 0.0918 (5)                       |
| O1   | -0.7131 (4)  | 0.4178 (4)   | -0.3468 (3)  | 0.0806 (10)                      |
| O2   | -0.6406 (5)  | 0.2829 (5)   | -0.4869 (4)  | 0.1119 (14)                      |
| O3   | -0.5192 (5)  | 0.0798 (5)   | -0.3027 (4)  | 0.1150 (14)                      |
| O4   | -0.7857 (7)  | -0.0410 (6)  | -0.4408 (5)  | 0.1371 (18)                      |
| O5   | -0.4458 (3)  | 0.3794 (3)   | -0.0203 (3)  | 0.0658 (9)                       |
| N1   | -0.3799 (5)  | 0.1982 (4)   | 0.0434 (4)   | 0.0631 (11)                      |
| N2   | -0.0880 (5)  | 0.2702 (5)   | 0.1614 (4)   | 0.0727 (11)                      |
| C1   | -0.8435 (9)  | 0.4438 (9)   | -0.4217 (6)  | 0.131 (3)                        |
| H1B  | -0.8231      | 0.5446       | -0.3730      | 0.196*                           |
| H1C  | -0.9517      | 0.3648       | -0.4345      | 0.196*                           |
| H1D  | -0.8443      | 0.4398       | -0.5086      | 0.196*                           |
| C2   | -0.6907 (10) | 0.2967 (8)   | -0.6060 (6)  | 0.140 (3)                        |
| H2B  | -0.6124      | 0.2972       | -0.6432      | 0.210*                           |
| H2C  | -0.6938      | 0.3925       | -0.5950      | 0.210*                           |
| H2D  | -0.8026      | 0.2105       | -0.6664      | 0.210*                           |
| C3   | -0.7194 (7)  | 0.2788 (5)   | -0.4025 (5)  | 0.0710 (14)                      |
| H3A  | -0.8400      | 0.2004       | -0.4578      | 0.085*                           |
| C4   | -0.5126 (8)  | -0.0639 (7)  | -0.3384 (6)  | 0.1142 (16)                      |
| H4A  | -0.3983      | -0.0432      | -0.2896      | 0.171*                           |
| H4B  | -0.5469      | -0.1044      | -0.4343      | 0.171*                           |
| H4C  | -0.5883      | -0.1389      | -0.3148      | 0.171*                           |
| C5   | -0.6598 (11) | 0.0736 (9)   | -0.3573 (8)  | 0.1142 (16)                      |
| C6   | -0.6513 (6)  | 0.2253 (6)   | -0.2981 (5)  | 0.0688 (13)                      |
| H6A  | -0.5298      | 0.3025       | -0.2460      | 0.083*                           |
| C7   | -0.7250 (6)  | 0.2132 (5)   | -0.1948 (5)  | 0.0517 (11)                      |
| C8   | -0.6183 (5)  | 0.2823 (4)   | -0.0593 (5)  | 0.0464 (10)                      |
| C9   | -0.6753 (6)  | 0.2753 (5)   | 0.0390 (5)   | 0.0609 (13)                      |
| H9A  | -0.5986      | 0.3259       | 0.1305       | 0.073*                           |
| C10  | -0.8481 (7)  | 0.1921 (6)   | 0.0000 (6)   | 0.0699 (14)                      |
| H10A | -0.8884      | 0.1850       | 0.0656       | 0.084*                           |
| C11  | -0.9602 (6)  | 0.1202 (6)   | -0.1340 (7)  | 0.0793 (16)                      |
| H11A | -1.0770      | 0.0644       | -0.1607      | 0.095*                           |
| C12  | -0.8968 (6)  | 0.1316 (5)   | -0.2302 (5)  | 0.0691 (14)                      |
| H12A | -0.9732      | 0.0823       | -0.3218      | 0.083*                           |
| C13  | -0.3283 (6)  | 0.3428 (5)   | 0.0417 (4)   | 0.0536 (11)                      |
| C14  | -0.2533 (6)  | 0.1731 (5)   | 0.1042 (5)   | 0.0720 (14)                      |
| H14A | -0.2858      | 0.0737       | 0.1068       | 0.086*                           |
| C15  | -0.0453 (6)  | 0.4133 (5)   | 0.1578 (4)   | 0.0586 (12)                      |
| C16  | -0.1620 (5)  | 0.4550 (5)   | 0.0963 (4)   | 0.0575 (12)                      |
| H16A | -0.1295      | 0.5538       | 0.0923       | 0.069*                           |

## supplementary materials

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### Atomic displacement parameters ( $\text{\AA}^2$ )

|     | $U^{11}$    | $U^{22}$    | $U^{33}$    | $U^{12}$    | $U^{13}$    | $U^{23}$    |
|-----|-------------|-------------|-------------|-------------|-------------|-------------|
| C1  | 0.0588 (8)  | 0.0939 (11) | 0.1023 (11) | 0.0264 (7)  | 0.0267 (8)  | 0.0357 (8)  |
| O1  | 0.104 (3)   | 0.067 (2)   | 0.066 (2)   | 0.047 (2)   | 0.023 (2)   | 0.0268 (18) |
| O2  | 0.161 (4)   | 0.177 (4)   | 0.075 (3)   | 0.115 (3)   | 0.076 (3)   | 0.085 (3)   |
| O3  | 0.118 (3)   | 0.119 (3)   | 0.127 (3)   | 0.084 (3)   | 0.050 (3)   | 0.033 (3)   |
| O4  | 0.137 (4)   | 0.122 (3)   | 0.136 (4)   | 0.083 (3)   | 0.037 (3)   | 0.013 (3)   |
| O5  | 0.0501 (19) | 0.0528 (18) | 0.097 (2)   | 0.0245 (16) | 0.0248 (17) | 0.0461 (18) |
| N1  | 0.065 (2)   | 0.042 (2)   | 0.092 (3)   | 0.0293 (19) | 0.037 (2)   | 0.035 (2)   |
| N2  | 0.064 (3)   | 0.073 (3)   | 0.099 (3)   | 0.039 (2)   | 0.039 (2)   | 0.050 (3)   |
| C1  | 0.169 (7)   | 0.195 (7)   | 0.096 (5)   | 0.143 (6)   | 0.064 (5)   | 0.062 (5)   |
| C2  | 0.239 (9)   | 0.150 (6)   | 0.093 (5)   | 0.133 (6)   | 0.082 (6)   | 0.067 (5)   |
| C3  | 0.100 (4)   | 0.053 (3)   | 0.064 (3)   | 0.036 (3)   | 0.039 (3)   | 0.028 (3)   |
| C4  | 0.127 (4)   | 0.126 (4)   | 0.125 (4)   | 0.092 (3)   | 0.056 (3)   | 0.047 (3)   |
| C5  | 0.127 (4)   | 0.126 (4)   | 0.125 (4)   | 0.092 (3)   | 0.056 (3)   | 0.047 (3)   |
| C6  | 0.084 (3)   | 0.078 (3)   | 0.058 (3)   | 0.046 (3)   | 0.034 (3)   | 0.032 (2)   |
| C7  | 0.059 (3)   | 0.044 (2)   | 0.060 (3)   | 0.029 (2)   | 0.027 (3)   | 0.027 (2)   |
| C8  | 0.051 (3)   | 0.038 (2)   | 0.057 (3)   | 0.027 (2)   | 0.020 (3)   | 0.025 (2)   |
| C9  | 0.090 (4)   | 0.045 (3)   | 0.053 (3)   | 0.037 (3)   | 0.030 (3)   | 0.024 (2)   |
| C10 | 0.085 (4)   | 0.061 (3)   | 0.104 (5)   | 0.044 (3)   | 0.063 (4)   | 0.053 (3)   |
| C11 | 0.048 (3)   | 0.063 (3)   | 0.120 (5)   | 0.021 (3)   | 0.032 (4)   | 0.044 (4)   |
| C12 | 0.068 (4)   | 0.054 (3)   | 0.065 (3)   | 0.023 (3)   | 0.019 (3)   | 0.014 (3)   |
| C13 | 0.062 (3)   | 0.048 (3)   | 0.062 (3)   | 0.032 (2)   | 0.031 (2)   | 0.025 (2)   |
| C14 | 0.068 (3)   | 0.053 (3)   | 0.113 (4)   | 0.039 (3)   | 0.041 (3)   | 0.044 (3)   |
| C15 | 0.062 (3)   | 0.059 (3)   | 0.060 (3)   | 0.025 (3)   | 0.036 (3)   | 0.027 (2)   |
| C16 | 0.053 (3)   | 0.048 (3)   | 0.068 (3)   | 0.021 (2)   | 0.027 (3)   | 0.026 (2)   |

### Geometric parameters ( $\text{\AA}$ , $^\circ$ )

|        |           |          |           |
|--------|-----------|----------|-----------|
| C1—C15 | 1.720 (5) | C3—H3A   | 0.9800    |
| O1—C1  | 1.410 (6) | C4—H4A   | 0.9600    |
| O1—C3  | 1.383 (5) | C4—H4B   | 0.9600    |
| O2—C2  | 1.318 (6) | C4—H4C   | 0.9600    |
| O2—C3  | 1.373 (5) | C5—C6    | 1.498 (8) |
| O3—C4  | 1.453 (6) | C6—C7    | 1.528 (6) |
| O3—C5  | 1.250 (7) | C6—H6A   | 0.9800    |
| O4—C5  | 1.186 (8) | C7—C8    | 1.363 (5) |
| O5—C8  | 1.393 (4) | C7—C12   | 1.376 (6) |
| O5—C13 | 1.341 (5) | C8—C9    | 1.370 (6) |
| N1—C13 | 1.343 (5) | C9—C10   | 1.378 (6) |
| N1—C14 | 1.327 (5) | C9—H9A   | 0.9300    |
| N2—C14 | 1.316 (5) | C10—C11  | 1.361 (7) |
| N2—C15 | 1.343 (5) | C10—H10A | 0.9300    |
| C1—H1B | 0.9600    | C11—C12  | 1.390 (7) |
| C1—H1C | 0.9600    | C11—H11A | 0.9300    |
| C1—H1D | 0.9600    | C12—H12A | 0.9300    |
| C2—H2B | 0.9600    | C13—C16  | 1.359 (5) |

|             |            |                |            |
|-------------|------------|----------------|------------|
| C2—H2C      | 0.9600     | C14—H14A       | 0.9300     |
| C2—H2D      | 0.9600     | C15—C16        | 1.370 (5)  |
| C3—C6       | 1.452 (6)  | C16—H16A       | 0.9300     |
| C3—O1—C1    | 117.8 (4)  | C5—C6—C7       | 108.8 (4)  |
| C2—O2—C3    | 126.5 (5)  | C3—C6—H6A      | 106.1      |
| C5—O3—C4    | 117.8 (5)  | C5—C6—H6A      | 106.1      |
| C13—O5—C8   | 121.1 (3)  | C7—C6—H6A      | 106.1      |
| C14—N1—C13  | 114.1 (4)  | C8—C7—C12      | 116.5 (4)  |
| C14—N2—C15  | 114.3 (4)  | C8—C7—C6       | 119.9 (4)  |
| O1—C1—H1B   | 109.5      | C12—C7—C6      | 123.6 (4)  |
| O1—C1—H1C   | 109.5      | C7—C8—C9       | 123.0 (4)  |
| H1B—C1—H1C  | 109.5      | C7—C8—O5       | 117.6 (4)  |
| O1—C1—H1D   | 109.5      | C9—C8—O5       | 118.9 (4)  |
| H1B—C1—H1D  | 109.5      | C8—C9—C10      | 118.9 (4)  |
| H1C—C1—H1D  | 109.5      | C8—C9—H9A      | 120.5      |
| O2—C2—H2B   | 109.5      | C10—C9—H9A     | 120.5      |
| O2—C2—H2C   | 109.5      | C11—C10—C9     | 120.4 (5)  |
| H2B—C2—H2C  | 109.5      | C11—C10—H10A   | 119.8      |
| O2—C2—H2D   | 109.5      | C9—C10—H10A    | 119.8      |
| H2B—C2—H2D  | 109.5      | C10—C11—C12    | 118.7 (5)  |
| H2C—C2—H2D  | 109.5      | C10—C11—H11A   | 120.6      |
| O2—C3—O1    | 114.2 (4)  | C12—C11—H11A   | 120.6      |
| O2—C3—C6    | 109.9 (4)  | C7—C12—C11     | 122.4 (5)  |
| O1—C3—C6    | 111.5 (4)  | C7—C12—H12A    | 118.8      |
| O2—C3—H3A   | 107.0      | C11—C12—H12A   | 118.8      |
| O1—C3—H3A   | 107.0      | O5—C13—N1      | 119.0 (4)  |
| C6—C3—H3A   | 107.0      | O5—C13—C16     | 117.2 (4)  |
| O3—C4—H4A   | 109.5      | N1—C13—C16     | 123.9 (4)  |
| O3—C4—H4B   | 109.5      | N2—C14—N1      | 128.5 (4)  |
| H4A—C4—H4B  | 109.5      | N2—C14—H14A    | 115.7      |
| O3—C4—H4C   | 109.5      | N1—C14—H14A    | 115.7      |
| H4A—C4—H4C  | 109.5      | N2—C15—C16     | 123.5 (4)  |
| H4B—C4—H4C  | 109.5      | N2—C15—C1      | 116.9 (4)  |
| O4—C5—O3    | 123.8 (7)  | C16—C15—C1     | 119.6 (4)  |
| O4—C5—C6    | 124.5 (7)  | C13—C16—C15    | 115.6 (4)  |
| O3—C5—C6    | 111.6 (7)  | C13—C16—H16A   | 122.2      |
| C3—C6—C5    | 112.0 (5)  | C15—C16—H16A   | 122.2      |
| C3—C6—C7    | 117.1 (4)  |                |            |
| C2—O2—C3—O1 | -67.6 (7)  | C13—O5—C8—C7   | -114.9 (4) |
| C2—O2—C3—C6 | 166.3 (6)  | C13—O5—C8—C9   | 72.8 (5)   |
| C1—O1—C3—O2 | 87.6 (6)   | C7—C8—C9—C10   | 1.4 (6)    |
| C1—O1—C3—C6 | -147.2 (5) | O5—C8—C9—C10   | 173.3 (3)  |
| C4—O3—C5—O4 | -3.6 (11)  | C8—C9—C10—C11  | -1.0 (6)   |
| C4—O3—C5—C6 | 173.0 (5)  | C9—C10—C11—C12 | 0.4 (7)    |
| O2—C3—C6—C5 | -53.2 (6)  | C8—C7—C12—C11  | 0.5 (6)    |
| O1—C3—C6—C5 | 179.1 (5)  | C6—C7—C12—C11  | 178.8 (4)  |
| O2—C3—C6—C7 | -179.9 (4) | C10—C11—C12—C7 | -0.1 (7)   |
| O1—C3—C6—C7 | 52.5 (6)   | C8—O5—C13—N1   | 12.4 (6)   |

## supplementary materials

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|              |            |                |            |
|--------------|------------|----------------|------------|
| O4—C5—C6—C3  | -54.2 (10) | C8—O5—C13—C16  | -169.3 (4) |
| O3—C5—C6—C3  | 129.3 (6)  | C14—N1—C13—O5  | 178.4 (4)  |
| O4—C5—C6—C7  | 76.8 (8)   | C14—N1—C13—C16 | 0.2 (6)    |
| O3—C5—C6—C7  | -99.7 (6)  | C15—N2—C14—N1  | 0.9 (8)    |
| C3—C6—C7—C8  | -128.5 (4) | C13—N1—C14—N2  | 0.0 (8)    |
| C5—C6—C7—C8  | 103.3 (5)  | C14—N2—C15—C16 | -2.0 (7)   |
| C3—C6—C7—C12 | 53.2 (6)   | C14—N2—C15—Cl  | 177.9 (4)  |
| C5—C6—C7—C12 | -75.0 (6)  | O5—C13—C16—C15 | -179.4 (4) |
| C12—C7—C8—C9 | -1.1 (6)   | N1—C13—C16—C15 | -1.2 (7)   |
| C6—C7—C8—C9  | -179.5 (4) | N2—C15—C16—C13 | 2.2 (7)    |
| C12—C7—C8—O5 | -173.1 (3) | Cl—C15—C16—C13 | -177.7 (3) |
| C6—C7—C8—O5  | 8.4 (5)    |                |            |

### Hydrogen-bond geometry ( $\text{\AA}$ , $^\circ$ )

| <i>D</i> —H $\cdots$ <i>A</i>      | <i>D</i> —H | H $\cdots$ <i>A</i> | <i>D</i> $\cdots$ <i>A</i> | <i>D</i> —H $\cdots$ <i>A</i> |
|------------------------------------|-------------|---------------------|----------------------------|-------------------------------|
| C6—H6A $\cdots$ O5                 | 0.98        | 2.25                | 2.777 (6)                  | 113                           |
| C1—H1B $\cdots$ Cg2 <sup>i</sup>   | 0.96        | 2.97                | 3.696 (5)                  | 134                           |
| C16—H16A $\cdots$ Cg1 <sup>i</sup> | 0.93        | 2.85                | 3.661 (4)                  | 146                           |

Symmetry codes: (i)  $-x+1, -y+1, -z$ .



Fig. 1

